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### Journal of Molecular Catalysis A: Chemical

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### Contents

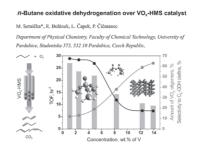
#### **Articles**

M. Setnička, R. Bulánek, L. Čapek, P. Čičmanec

Journal of Molecular Catalysis A: Chemical 344 (2011) 1

n-Butane oxidative dehydrogenation over  $VO_{\chi}$ -HMS catalys

► VO<sub>x</sub> units on mesoporous HMS were studied by means of analytic techniques. ► Monomeric VO<sub>x</sub> units play role of most active and selective site in ODH of n-butane. ► Amount of monomeric units is comparable for synthesized and impregnated samples. ► Presence of O<sub>h</sub>-oligomeric units causes decreasing of selectivity to C4-ODH products. ► Higher selectivity to C4-ODH products exhibit samples prepared by direct synthesis.

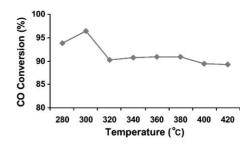


#### Ali Reza Salehi Rad, Maryam behzad khoshgouei, Ali Reza Rezvani

Journal of Molecular Catalysis A: Chemical 344 (2011) 11

Water gas shift reaction over  $Zn-Ni/SiO_2$  catalyst prepared from  $[Zn(H_2O)_6]_2[Ni(NCS)_6]\cdot H_2O/SiO_2$  precursor

▶ The method of Zn–Ni/SiO $_2$  catalyst preparation is a simple and suitable way. ▶ The Zn–Ni catalyst has the high catalytic activity for WGS reaction at 280–420 °C. This catalyst presents higher activity than those prepared from other methods.



## Mannar R. Maurya, Manisha Bisht, Fernando Avecilla

Journal of Molecular Catalysis A: Chemical 344 (2011) 18

Synthesis, characterization and catalytic activities of vanadium complexes containing ONN donor ligand derived from 2-aminoethylpyridine

▶ Synthesis and characterization of oxidovanadium(IV and V) complexes with new ligand derived from pyridoxal and 2-aminoethylpyridine (Hpydx-aepy). ▶ Structure of [V<sup>IV</sup>O(acac)(pydx-aepy)] (1) has been solved by single crystal X-ray. ▶ Formation of the peroxido complex in solution has also been monitored by electronic absorption spectroscopy. ▶ Encapsulation of [V<sup>V</sup>O<sub>2</sub>(pydx-aepy)] in the cavity of zeolite-Y and their catalytic activity for the oxidation of styrene, methyl phenyl sulfide, diphenyl sulfide and cyclohexene. ▶ Catalytic results are very good.

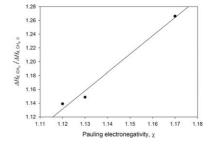
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#### Say Yei Foo, Chin Kui Cheng, Tuan-Huy Nguyen, Adesoji A. Adesina

Journal of Molecular Catalysis A: Chemical 344 (2011) 28

Evaluation of lanthanide-group promoters on Co–Ni/Al<sub>2</sub>O<sub>3</sub> catalysts for CH<sub>4</sub> dry reforming

▶ Lanthanide doping did not appear to affect  $CH_4$  and  $CO_2$  consumption rates. ▶ However, rare-earth promotion increased  $H_2$  and CO production rates. ▶ Carbon deposition on the promoted catalysts reduced by up to 50%. ▶ Catalyst attributes correlated well with Pauling electronegativity of the dopants.

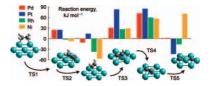


#### Duygu Basaran, Hristiyan A. Aleksandrov, Zhao-Xu Chen, Zhi-Jian Zhao, Notker Rösch

Journal of Molecular Catalysis A: Chemical 344 (2011) 37

Decomposition of ethylene on transition metal surfaces M(1 1 1). A comparative DFT study of model reactions for M = Pd, Pt, Rh, Ni

▶ Theoretical study of ethylene decomposition on M(1 1 1) surfaces, M = Pd, Pt, Rh, Ni. ▶ Species with more H atoms dehydrogenate more easily than species with fewer H. ▶ Dehydrogenation occurs easier on Ni(1 1 1) and Rh(1 1 1) than on Pd(1 1 1) and Pt(1 1 1). ▶ Reactivity of Pd (Rh) regarding ethylene decomposition similar to that of Pt (Ni). ▶ and Rh favor  $C_2$  decomposition whereas on Ni  $C_2$  formation is favored.

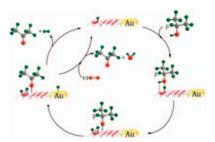


# Z. Martinez-Ramirez, S.A. Jimenez-Lam, J.C. Fierro-Gonzalez

Journal of Molecular Catalysis A: Chemical 344 (2011) 47

Infrared spectroscopic evidence of adsorbed species during the oxidation of 2-propanol catalyzed by  $\gamma\text{-Al}_2O_3\text{-supported gold: Role of gold as a hydrogen-subtractor$ 

▶  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>-supported gold is catalytically active for the oxidation of 2-propanol. ▶ IR spectra of functioning catalysts identified species bonded to the support. ▶ Results suggest that the alcohol is activated on the support. ▶ Role of gold consists of subtracting hydrogen from  $\beta$ -C-H bond of surface alkoxide.



### Xiao-Xiang He, Chen Fan, Xiong-Yi Gu, Xing-Gui Zhou, De Chen, Yi-An Zhu

Journal of Molecular Catalysis A: Chemical 344 (2011) 53

Role of  $CO_2$  in ethylbenzene dehydrogenation over  $Fe_2O_3(0\ 0\ 0\ 1)$  from first principles

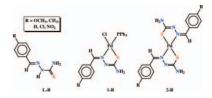
▶ The mechanism for ethylbenzene dehydrogenation in the presence of  $CO_2$  is explored. ▶ Styrene is hard to escape from the most active O-terminated  $Fe_2O_3(0\ 0\ 0\ 1)$ . ▶ The Fe-terminated surface dominates the reaction, with the coupling mechanism. ▶ Both the one-step and two-step pathways are probable while the former is dominant.

Piyali Paul, Sayanti Datta, Sarmistha Halder, Rama Acharyya, Falguni Basuli, Ray J. Butcher, Shie-Ming Peng, Gene-Hsiang Lee, Alfonso Castineiras, Michael G.B. Drew, Samaresh Bhattacharya

Journal of Molecular Catalysis A: Chemical 344 (2011) 62

Syntheses, structures and efficient catalysis for C–C coupling of some benzaldehyde thiosemicarbazone complexes of palladium

► Reaction of 4-R-benzaldehyde thiosemicarbazones with [Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>] afford complexes (**1-R**) containing a thiosemicarbazone, a PPh<sub>3</sub> and a chloride. ► Similar reaction with Na<sub>2</sub>[PdCl<sub>4</sub>] afford bis-thiosemicarbazone complexes (**2-R**). ► Coordination to Pd is associated with a conformational change around the C=N Bond ► Both **1-R** and **2-R** complexes can efficiently catalyze C-C coupling reactions.



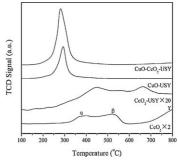
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#### Qinqin Huang, Xiaomin Xue, Renxian Zhou

Journal of Molecular Catalysis A: Chemical 344 (2011) 74

Catalytic behavior and durability of  $CeO_2$  or/and CuO modified USY zeolite catalysts for decomposition of chlorinated volatile organic compounds

▶ The catalytic activity for CVOCs destruction is evidently enhanced over modified USY catalysts. ▶ The high activity is due to high dispersion of  $CeO_2$  or CuO, good oxygen mobility and Lewis acidity. ▶ Modified USY catalysts present high selectivity to HCl and  $CO_2$  formation. ▶ Interaction between CuO and  $CeO_2$  improves the durability of the catalyst in long term reaction.



#### Savita Khare, Rajendra Chokhare

Journal of Molecular Catalysis A: Chemical 344 (2011) 83

Synthesis, characterization and catalytic activity of Fe(Salen) intercalated  $\alpha\mbox{-}zirconium$  phosphate for the oxidation of cyclohexene

▶ Synthesis of a heterogeneous catalyst,  $\alpha$ -ZrP·Fe(Salen) by flexible ligand method. ▶ Catalyst characterized by BET, XRD, SEM, EDX, FTIR, AAS and Mössbauer spectroscopy. ▶ Catalytic activation of  $\alpha$ -ZrP·Fe(Salen) with dry TBHP for oxidation of cyclohexene. ▶ Study of recycling of the catalyst up to eight cycles.

### R.M. Hassan, S.M. Ibrahim, I.A. Zaafarany, A. Fawzy, H.D. Takagi

Journal of Molecular Catalysis A: Chemical 344 (2011) 93

Base-catalyzed oxidation: Kinetics and mechanism of hexacyanoferrate (III) oxidation of methyl cellulose polysaccharide in alkaline solutions

► A kinetic study of the oxidation of some natural polymeric compounds such as methyl cellulose polysaccharides by alkaline ferricyanide (III). ► A novel synthesis of diketo-derivatives of methyl cellulose by an oxidation method. ► Examining the behavior of polysaccharides containing alcoholic groups in aqueous alkaline solutions. ► Elucidation of reaction mechanism for the oxidation process of the cited work.

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#### Ruibin Jiang, Wenyue Guo, Ming Li, Houyu Zhu, Lianming Zhao, Xiaqing Lu, Honghong Shan

Journal of Molecular Catalysis A: Chemical 344 (2011) 99

Methanol dehydrogenation on Rh(1 1 1): A density functional and microkinetic modeling study

▶ Rh(1 1 1)-catalyzed methanol dehydrogenation is studied using theory modeling. ▶ The reaction mechanism is identified under two different reaction conditions. ▶ The reason why oxidation does not take place at  $CH_2O$  in methanol oxidation is found. ▶ The origin of different mechanisms of the reaction on different VIII metals is found.

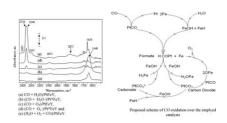
# Zeinhom M. El-Bahy, Ahmed I. Hanafy, Mohamed M. Ibrahim, Masakazu Anpo

Journal of Molecular Catalysis A: Chemical 344 (2011) 111

In situ FTIR studies of CO oxidation over Fe-free and Fe-promoted PtY catalysts: Effect of water vapor addition

▶ Preparation and characterization of PtY and PtFeY catalysts by ion exchange method. ▶ In situ FTIR studies of CO oxidation, WGS over as prepared and reduced catalysts. ▶ Study the effect of addition of trace amount of water on the oxidation of CO with  $O_2$ . ▶ Addition of  $H_2O$  and/or  $H_2O + O_2$  enhanced CO removal over Fe-free and Fe-promoted PtY. ▶ Admission of  $(CO + O_2 + H_2O)$  mixture increased the adsorbed amount of  $CO_2$  over PtFeY.

Promotion of PtY with Fe oxide and addition of H<sub>2</sub>O with O<sub>2</sub> to the surface of the



### Rong Wang, Yonghong Li, Ronghui Shi, Meimei Yang

Journal of Molecular Catalysis A: Chemical 344 (2011) 122

Effect of metal–support interaction on the catalytic performance of Ni/Al $_2$ O $_3$  for selective hydrogenation of isoprene

► The different metal–support interactions over two alumina supports were characterized. ► The effect of the different interaction on catalytic performance was explained. ► The weak interaction resisting coke deposition was related to the hydrogenolytic sites.

### Alireza Khorshidi, Khalil Tabatabaeian

Journal of Molecular Catalysis A: Chemical 344 (2011) 128

Ruthenium-exchanged FAU-Y zeolite catalyzed improvement in the synthesis of 6*H*-indolo[2,3-*b*]quinolines

► A convenient method for preparation of indoloquinolines is reported. ► RuY as a heterogeneous catalyst resulted in more efficiency. ► Reusability of the solid acid catalyst is also, noticeable.

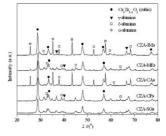
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#### Qiuyan Wang, Zhenguo Li, Bo Zhao, Guangfeng Li, Renxian Zhou

Journal of Molecular Catalysis A: Chemical 344 (2011) 132

Effect of synthesis method on the properties of ceria–zirconia modified alumina and the catalytic performance of its supported Pd-only three-way catalyst

► The ceria–zirconia modified alumina (CZA) was prepared by five different methods. ► The effect of preparation methods on the structural properties of CZA was studied. ► Coprecipitation with supercritical drying leads to good thermal stability of CZA. ► The corresponding Pd-only three-way catalyst exhibits higher catalytic performance.

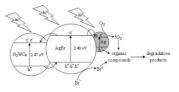


#### Jing Cao, Bangde Luo, Haili Lin, Shifu Chen

Journal of Molecular Catalysis A: Chemical 344 (2011) 138

Synthesis, characterization and photocatalytic activity of AgBr/H<sub>2</sub>WO<sub>4</sub> composite photocatalyst

▶ AgBr/ $H_2WO_4$  was synthesized by using a facile deposition–precipitation method. ▶ AgBr/ $H_2WO_4$  displays excellent visible-light photocatalytic activity ( $\lambda > 420 \text{ nm}$ ). ▶ AgBr/ $H_2WO_4$  possesses good stability after successive 5 cycle experiments. ▶ The resulting  ${}^{\bullet}O_2^{\phantom{0}}$ , Br $^0$  and h $^+$  played the major roles for MO and RhB degradation.

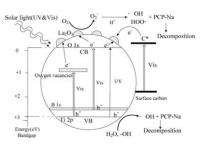


### J.W. Liu, R. Han, H.T. Wang, Y. Zhao, W.J. Lu, H.Y. Wu, T.F. Yu, Y.X. Zhang

Journal of Molecular Catalysis A: Chemical 344 (2011) 145

Degradation of PCP-Na with La-B co-doped  ${\rm TiO}_2$  series synthesized by the sol-gel hydrothermal method under visible and solar light irradiation

► Novel La-B-TiO<sub>2</sub> synthesized by sol-gel hydrothermal route with high activity. ► Efficient decomposition and dechlorination of PCP-Na under visible and sun light. ► Synergetic effects of La and B implantation. ► Variation of catalytic activity with the action of the dopants in modified system.



### Jo-Yong Park, Yun-Jo Lee, Prashant R. Karandikar, Ki-Won Jun, Jong Wook Bae, Kyoung-Su Ha

Journal of Molecular Catalysis A: Chemical 344 (2011) 153

Ru promoted cobalt catalyst on  $\gamma\text{-}Al_2O_3$  support: Influence of pre-synthesized nanoparticles on Fischer–Tropsch reaction

► Controlled size  $CoRuO_x$  nanoparticles were embedded on  $\gamma$ - $Al_2O_3$ . ► Intimate contact between Ru and Co increased by pre-synthesis of nanoparticles. ► 5CoxRuAl catalysts show superior activity in Fischer–Tropsch reaction. ► Conventional catalyst with equal amount of Co and Ru show comparatively poor result. ► Increased interaction of

Ru and Co in 5CoxRuAl enhances catalyst activity.

